

1 **Effect of the operating conditions on the anaerobic digestion of wheatgrass for**
2 **chemicals and energy production**

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6 **Abstract**

7 The aim of this study was to investigate anaerobic digestion of wheatgrass in the
8 absence of hydrolytic pre-treatments. The effect of solids retention time (SRT) (1-64
9 d), inoculum acclimation (0-80 d acclimation), temperature (40-70 °C) and buffer
10 capacity (20-200 mM phosphate buffer) on conversion of the feedstock, yield and
11 composition of liquid-phase products (ethanol and short-chain organic acids, SCOAs)
12 and COD removal was investigated in semi-continuous (intermittent feed) completely
13 mixed reactors.

14 SRT had the most important effect on process performance. Biodegradation of the
15 feedstock was favoured at high SRT, with 61 % removal of volatile suspended solids
16 and 84 % removal of total carbohydrates at SRT 64 d. However, low yield of liquid-
17 phase products was observed at high SRT because of strong methanogenic activity
18 (57 % removal of the total COD). The highest yield of liquid-phase products was 20 %
19 (COD basis) at SRT 8 d. Although high biodegradation of the feedstock was observed
20 after long-term batch acclimation (30 and 80 d), once the digestion conditions were
21 switched to semi-continuous at short SRT (2 d) the biodegradation of the feedstock
22 decreased considerably. The best process performance was observed at 40 °C.

23 Keywords: anaerobic digestion, carboxylates, lignocellulose, grass, biorefinery

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25 **Introduction**

26 Anaerobic digestion (AD) of biomass, often organic waste, is typically used across the
27 globe for the generation of methane to be converted into energy. In recent years, the
28 possible use of AD in a biorefinery context to produce bio-based products such as
29 short chain organic acids (SCOAs) and hydrogen is becoming of interest [1,2]. The
30 use of organic waste to produce these chemicals is attractive because of the non-
31 renewable nature of the feedstocks often used in current commercial processes.

32 According to our recent study [3], on a global scale the current energy generation by
33 anaerobic digestion is only a minor fraction of the total potential energy that could be
34 generated based on the generation rate of organic waste. One of the limitations in the
35 use of anaerobic digestion for energy or chemicals production is the lignocellulosic
36 nature of many types of organic waste (e.g. agricultural residues). Lignocellulosic
37 biomass, characterised by the presence of lignin, cellulose and hemicellulose, is
38 typically slowly biodegradable and this has so far limited its use as feedstock for
39 anaerobic digesters.

40 Many literature studies on the use of lignocellulosic biomass for fermentation purposes
41 performed pre-treatments of the feedstock before the fermentation stage [4]. The aim
42 of the pre-treatments is to hydrolyse the lignin and cellulosic components of biomass,
43 converting the cellulosic components into sugars that can be easily digested. Pre-
44 treatments usually require alkaline or acidic solutions, enzymes (e.g. cellulase,
45 cellobiase and xylase) and/or energy [5, 6]. Although the pre-treatments usually give
46 good yields of carbohydrates released or converted from the feedstock, the costs
47 associated with chemicals, enzymes and/or energy in the pre-treatment and hydrolysis
48 steps are expensive for industrial scale.

49 As an alternative to hydrolytic pre-treatments, microorganisms in AD processes are in
50 principle able to develop the enzymes required to hydrolyse lignin and cellulose.

51 Therefore, an AD process of lignocellulosic biomass without hydrolytic pre-treatments
52 is, at least in theory, possible. However, little investigation has been reported in the
53 literature on the use of lignocellulosic biomass for AD without hydrolytic pre-
54 treatments.

55 The novelty and aim of this study is the experimental investigation of the AD of
56 lignocellulosic substrates without hydrolytic pre-treatments. As an example of
57 lignocellulosic biomass, we used wheatgrass, in particular the young shoots of the
58 crop *Triticum aestivum* (common wheat). The lignin, cellulose and hemicellulose
59 content of wheatgrass and of other grasses depends on the species, on the growth
60 conditions and on the age of the grass at harvest [7,8]. For example, the lignin content
61 of young (27 d) wheat plants was in the range 4-8 % [9]. The neutral detergent fibre
62 (approximately corresponding to the sum of cellulose, hemicellulose and lignin) and
63 the acid detergent fibre (approximately corresponding to the sum of cellulose and
64 lignin) were 15-17 % and 5-7 %, respectively, in several varieties of wheatgrass
65 sprouts at 5 d age [10]. Wheatgrass belongs to the family of Gramineae, which can
66 grow in non-arable lands without direct competition with food production or
67 requirement of fertilisers or pesticides. Owing to its widespread growth on land,
68 Gramineae have been considered as a potential lignocellulosic feedstock for
69 biochemicals and bioenergy production [11]. The substrate used in this study had only
70 been subject by the manufacturer to the pre-treatments of particle reduction (milling)
71 and drying, which do not alter the chemical structure of the feedstock.

72 Our study investigated the effect of operating conditions (residence time, acclimation
73 of the inoculum, temperature and pH), on substrate conversion, yield and composition
74 of liquid phase products and COD removal in semi-continuous (intermittent feed)
75 experiments.

76

77 **Material and methods**

78 *Feedstock of the reactors*

79 Wheatgrass powder obtained from sprouts of the *Triticum aestivum* L. (Bulk Powders,
80 UK) was used as the lignocellulosic substrate for the fermentation process. The
81 substrate was produced from raw wheatgrass after drying and grinding and was
82 comprised of a fine powder (200 mesh) of 100 % wheatgrass with 50 % of
83 carbohydrates, 25 % of proteins and 25 % of fibre [12]. The feed of the reactors
84 consisted in a suspension of 20 g/l of wheatgrass in tap water. The suspension also
85 contained the following mineral salts: K_2HPO_4 , 17.4 g/l; NaH_2PO_4 , 12.0 g/l; NH_4Cl , 2
86 g/l; $MgCl_2 \cdot 6H_2O$, 0.125 g/l; $CaCl_2 \cdot H_2O$, 0.09 g/l. This suspension corresponded to a
87 concentration of phosphate buffer of 200 mM and Table 1 presents the
88 characterisation of the feedstock suspension. In the run with low concentration of
89 phosphate buffer (20 mM, Run 13 in Table 2), the concentration of K_2HPO_4 and
90 NaH_2PO_4 was 1.74 g/l and 1.2 g/l, respectively.

91 *Experimental setup*

92 A summary of the experimental conditions in all the runs is reported in Table 2. The
93 scheme of the reactor set-up is shown in Figure 1. Stirred (magnetic bar at 250 rpm)
94 jacketed glass vessels of 200 ml working volume (300 ml total volume) were used as
95 fermenters. The circulating water in the jacket was provided by a heating circulator
96 and used to control the fermentation broth temperature at the desired value. A
97 thermostat was dipped into the fermentation broth to measure the temperature. A
98 silicone tubing was connected to the fermentation broth for sampling with a volumetric
99 pipette. The effluent was collected continuously with an overflow tubing set to maintain
100 the working volume at 200 ml and the outlet silicone tubing was bent in a U shape to
101 prevent oxygen entrance and maintain anaerobic conditions inside the fermenter. The

102 system was well mixed and the SRT (solids retention time) coincided with the HRT
103 (hydraulic retention time).

104 The wheatgrass suspension used as feed for the reactors was also continuously
105 stirred with a magnetic bar at 250 rpm. A peristaltic pump was used for feeding. The
106 pump had been previously calibrated at flowrate of 25 ml/min and was used to pump
107 intermittently the feed suspension to the fermenter. The pump was switched on for
108 one minute via a programmable power management system, so that each feeding
109 cycle corresponded to feeding of 25 ml into each reactor. The flowrate and feeding
110 time were maintained constant in all the runs to preserve the same flow regime and
111 residence time of particles in the pumping system, respectively.

112 Each experimental run was initiated by filling the fermenter with fresh feedstock
113 suspension and subsequently sealing the vessel from the atmospheric air with a multi-
114 port lid of polytetrafluoroethylene. Nitrogen gas (oxygen-free) was used to sparge for
115 5 minutes the suspension in the fermenter and favour an anaerobic condition inside
116 the vessel. The inoculum was taken from the digestate of a commercial anaerobic
117 digester (Gask Farm, Turriff, Aberdeenshire, UK). The digester treats food industry
118 and agricultural waste and operates at 38-40 °C. The inoculum concentration used to
119 start-up the reactor was 1 g VSS/l. The inoculum was stored in fridge (2 °C) in order
120 to maintain freshness and minimise its degradation and was maintained at room
121 temperature for 24 h before use, similarly to other literature studies [13]. The inoculum
122 was used without acclimation to the feedstock, except for the runs (Runs 8 and 9)
123 which investigated the effect of acclimation. In these runs, the inoculum was
124 acclimated to the feed suspension in batch for the specified length of time, before the
125 cultivation conditions were switched to semi-continuous.

126 The length of the fermentation runs was at least 3.5 times the SRT, with a minimum
127 of 40 d, in order to obtain enough data to calculate the average values at steady state

128 of the measured variables. The average values for each run were calculated from all
129 the samples collected during the run, ignoring the samples collected during the first
130 three SRTs, which was considered the time necessary to reach the steady state.

131 *Analytical procedures*

132 The reactors were sampled once or twice per week. At each sample day, the
133 fermentation broth was sampled to measure the concentration of VSS, TC,
134 fermentation products, TCOD and the pH.

135 An aliquot of 5 ml of the fermentation broth was sampled with a volumetric pipette to
136 filter in a pre-weighted glass-fibre filter paper to determine the VSS concentration.
137 Aliquots samples of 1 μ l of the filtrate broth were used to determine the concentration
138 of the fermentation products (acetic, propionic and butyric acid and ethanol) on GC-
139 FID using a capillary column (30m x 0.25mm, Thermo Scientific TraceGOLD™ TG-
140 WaxMS A). The initial temperature of the column was 80 °C for 2 min followed with a
141 ramp of 20 °C/min and a final temperature of 200 °C for 1 min; the injector and detector
142 temperatures were 200 °C and 250 °C respectively. Hydrogen was used as the carrier
143 gas at a flowrate of 35 ml/min. The samples were acidified with H₃PO₄ (30% v/v) and
144 2-ethyl-butyric acid was used as internal standard. Total and soluble carbohydrates
145 were measured according to the Anthrone method. Total and soluble COD (TCOD
146 and SCOD) were measured in Spectroquant® COD cell test (Merck COD kits,
147 Germany). An aliquot of 3 ml of the fermentation broth was used to measure the pH
148 (pH meter Metler Toledo, Switzerland).

149 *Calculations*

150 The COD conversion factors for the fermentation products were calculated according
151 to their respective oxidation stoichiometries. The removal of VSS, TC and TCOD were
152 calculated according to Equations (1-3) :

$$VSS \text{ removal (\%)} = \left(1 - \frac{VSS \text{ in the fermentate } \left(\frac{g}{l}\right)}{VSS \text{ in the feed } \left(\frac{g}{l}\right)}\right) \times 100\% \quad (1)$$

$$TC \text{ removal (\%)} = \left(1 - \frac{TC \text{ in the fermentate } \left(\frac{g}{l}\right)}{TC \text{ in the feed } \left(\frac{g}{l}\right)}\right) \times 100\% \quad (2)$$

$$TCOD \text{ removal (\%)} = \left(1 - \frac{TCOD \text{ in the fermentate } \left(g \frac{COD}{l}\right)}{TCOD \text{ in the feed } \left(g \frac{COD}{l}\right)}\right) \times 100\% \quad (3)$$

153 The yield of each liquid phase product was calculated by dividing the concentration of
 154 each substance, in COD units, by the total COD of the feed. The total products yield
 155 was calculated by adding together the concentrations of all the liquid-phase products
 156 (in COD units) and dividing by the total COD of the feed (Equation (4)).

Total products yield (%)

$$= \frac{\sum \text{Products in the fermentate } (g \text{ COD}/l)}{\text{Feed concentration } (g \text{ COD}/l)} \times 100\% \quad (4)$$

157 The methane yield on the VS (volatile solids) of the feed was calculated for Run 7
 158 (SRT 64 d) at steady state because in this run a high removal of the TCOD was
 159 observed. The methane yield was calculated from the COD balance, which states that
 160 under anaerobic conditions any removal of TCOD from the liquid-solid phase is due to
 161 COD leaving the system with the gas phase [14]. The species that can account for the
 162 COD leaving with the gas phase are hydrogen and methane, but in the conditions of
 163 Run 7 it is reasonable to assume that methane and not hydrogen was the main product
 164 (this will be discussed in the *Results and discussion, Effect of the SRT* section). The
 165 steps in the calculation of the methane yield were the following:

- 166 - the TCOD of the feed was multiplied by the average TCOD removal at steady
 167 state to obtain the average mass of TCOD removed per unit volume of the
 168 reactor (kg COD/m³);

- 169 - The mass of TCOD removed per unit volume of the reactor was divided by the
170 COD conversion factor for methane (4 kg COD/kg CH₄, calculated from the
171 stoichiometry of total oxidation of methane) to obtain the methane production
172 per unit volume of the reactor (kg CH₄/m³);
- 173 - The methane production per unit volume of the reactor was converted into
174 volume (m³ CH₄/m³) using the ideal gas law at the temperature of the reactor
175 (40 °C), which gives a methane density of 0.62 kg/m³;
- 176 - The volumetric methane production per unit volume of the reactor was divided
177 by the VS concentration of the feed to obtain the methane production per unit
178 of VS (m³ CH₄/kg VS).

179

180 **Results and discussion**

181 *Effect of SRT*

182 The effect of SRT was investigated in Runs 1-7. The time profiles of VSS and TC
183 removal and of the total yield of products are shown in Figure 2, while the average
184 values at steady state of VSS, TC and TCOD removal from the liquid-solid phase are
185 shown in Figure 3. The general trend shows an increase in VSS, TC and TCOD
186 removal with increasing SRTs. Whilst the VSS reduced only 9 % at SRT of 1 d, a
187 reduction of over 60 % was observed in the VSS for the SRT at 64 d. Regarding TC
188 removal, whilst approximately 15 % of TC concentration reduced for the fermenter at
189 SRT of 1 d, over 84 % of TC reduction was observed at SRT of 64 d. No TCOD removal
190 from the liquid phase was observed for SRT 1-4 d. TCOD removal was only observed
191 for SRT values of 8 d or higher and reached the highest value of 57 % at 64 d of SRT.
192 The total products yield was between 10 and 20 % (COD basis) for runs at SRT 1-8 d
193 and SRT 32 d, while it was virtually negligible in the runs at SRT 16 and 64 d. The
194 maximum yield of total products was 20 % (COD basis) for SRT 8 d. In all runs the
195 removal of soluble carbohydrates was complete after two days of start-up (data not
196 shown).

197 The general evidence from Figure 3 is that in all runs some hydrolysis of the insoluble
198 material in the feed was observed, as indicated by the removal of VSS. It should be
199 noted that microorganisms generated in the fermentation contribute to the VSS, which
200 complicates the analysis of VSS removal. However, the production of microorganisms
201 in anaerobic digestion is low (for example the study by Kalyuzhnyi [15] reports
202 literature and own values for anaerobic growth yields, the highest value being 0.12 g/g
203 of glucose for acidogenic microorganisms) and therefore VSS removal can be used
204 for a qualitative indication of the removal of the insoluble organic matter in the feed. In
205 most runs, removal of insoluble carbohydrates was also observed, as indicated by the

206 fact that the removal of SC corresponded to about 17% of the total carbohydrates in
207 the feed, and the removal of TC was, in all runs except the one at SRT 1 d, higher
208 than this value. Acetate was the main component in the fermentation products,
209 followed by propionate and butyrate. Throughout this study, the composition of the
210 products remained relatively constant and unaffected by the process conditions, and
211 also remained relatively constant within each run.

212 The generation of gas-phase products (hydrogen and/or methane) was estimated,
213 using the COD balance (see *Materials and methods, Calculations* section), from the
214 removal of TCOD from the liquid-solid phase. The use of the COD balance has been
215 shown to give accurate estimations of methane production in methanogenic anaerobic
216 digesters in many experimental studies (e.g., [16, 17]). Thermodynamic
217 considerations (not shown) on the gas-liquid equilibrium of the other products (short
218 chain organic acids and ethanol) allow to exclude any significant losses of these
219 substances with the gas phase. From Figure 3 the general evidence is that for SRT 1-
220 4 d all the fermentation products were present in the liquid-solid phase (no reduction
221 in the TCOD of the liquid-solid phase), while at longer SRT the generation of gas-
222 phase products gradually increased in importance. Gas-phase products became the
223 main fermentation products at the longest investigated SRT, 64 d, where none of the
224 analysed liquid-phase products (acetic, propionic, butyric and ethanol) were detected
225 for most of the run. Although our experimental set-up did not distinguish between
226 hydrogen and methane, the high COD removal at SRT 64 d is clearly an indication
227 that methanogenesis was taking place, for the following reasons. Firstly, hydrogen
228 generation from the fermentation of organic species can only account for up to 30-35
229 % of the COD of the feedstock [1] whereas TCOD removal at SRT 64 d was 57 %.
230 The 30-35 % COD conversion into hydrogen is a theoretical maximum and is usually
231 not achieved under anaerobic digestion conditions, unless the process conditions are

232 carefully selected to prevent the growth of hydrogen consuming bacteria (e.g. pre-
233 treatment of the inoculum or low pH). Secondly, unless the growth of hydrogen
234 consuming bacteria is inhibited by other process conditions (as just described),
235 hydrogen production from AD is usually associated with short SRT, typically less than
236 1 d [18], while in this case the SRT was much longer and typical of methanogenic
237 conditions. Thirdly, virtually no removal of TCOD was observed in the runs at the
238 shortest SRTs (1-4 d), where hydrogen generation was expected. This indicates that
239 with the composition of the considered feedstock, hydrogen producing and hydrogen
240 consuming reactions were balanced, with no net production of hydrogen under any
241 conditions. Assuming, as justified above, that the 57 % reduction in TCOD of the liquid
242 phase obtained at SRT 64 d is entirely due to methane production, the methane yield
243 at this SRT corresponds to 0.29 m³ CH₄/kg VS.

244 The runs at the three longest values of SRT showed a non-monotonic trend in the
245 products yield (Figure 3b), with negligible yield at SRT 16 and 64 d and higher product
246 yield at SRT 32 d. The reason for this behaviour is not clear although it can be
247 hypothesised that SRT 16-32 d is a borderline residence time between the conditions
248 of full acidogenesis (observed for runs at SRT up to 8 d) and full methanogenesis
249 (observed at SRT 64 d). The run at SRT 32 d showed alternation of periods of high
250 and low products yield, indicating instability, while the run at SRT 16 d showed the
251 absence of liquid phase products throughout most of the operating period (Figure 2i).
252 Possibly, in the intermediate range of SRT the process behaviour can shift from the
253 acidogenic to the methanogenic behaviour due to small changes in the process
254 conditions, for example in the activity of the microbial populations in the inoculum.
255 However, with the experimental data collected in this study, we have no proof for this
256 hypothesis and the behaviour of the process at intermediate values of the SRT
257 deserves further investigation.

258 The run at the longest SRT (64 d) showed an interesting profile for the VSS and TC
259 removal (Figures 2c and 2f). The removal of VSS and TC increased gradually from the
260 start of the run until approximately day 70, when a sharp increase in their removal was
261 observed. These profiles indicate the slow phenomena of growth and acclimation on
262 the lignocellulosic substrate. In this run, the concentration of total products (Figure 2i)
263 was high in the initial days after start-up, then decreased virtually to zero, indicating
264 acclimation and growth of methanogenic microorganisms. However, in coincidence
265 with the step increase in VSS and TC removal, the concentration of total products also
266 showed a step increase, but it came back to negligible values within a few days. These
267 profiles indicate that the increase in the hydrolysis and acidogenesis rate of the
268 microbial culture was not initially matched by the methanogenesis rate, but in a few
269 days methanogenic microorganisms were able to grow fast enough and to be able to
270 metabolise the acids to methane at an increased rate. A similar sharp increase in
271 methane generation was also observed by Turick *et al.* [19] during the batch anaerobic
272 digestion of woody biomass, after more than 50 d of digestion. In that case, however,
273 the products in the liquid phase were not measured.

274 In general, the boundary between acidogenic and methanogenic activity was around
275 of 8-32 d of SRT. This shift between acidogenic and methanogenic microorganisms
276 varying the SRT was also observed by several researchers for grass [20] and other
277 organic substrates [21]. Since methanogens are slow growers, a minimum SRT is
278 required to observe methane production for a particular feedstock. Since these studies
279 aimed at biomethane production, the production of SCOAs and drop in the pH at low
280 SRTs was considered a fermentation failure.

281 To further illustrate the mass balances, Figure 4 shows the fate of the total COD of the
282 feed in Runs 1-7. When the COD conversion was mainly directed to liquid-phase
283 products, the maximum conversion of the feed COD was 22 % (runs at SRT 8 and 32

284 d), with 78 % of the feed COD being either unconverted or converted into
285 microorganisms. When the COD conversion was directed to methane, the majority of
286 the feed COD was converted (run at SRT 64 d).

287 The highest reduction in TCOD obtained in this study, 57 %, which corresponds to
288 $0.29 \text{ m}^3 \text{ CH}_4/\text{kg VS}$, indicates one of the highest methane productions reported in the
289 literature for lignocellulosic substrates without hydrolytic pre-treatments. Tong *et al.*
290 [22] reported up to $0.36 \text{ m}^3 \text{ CH}_4/\text{kg VS}$ for corn stover (10 % lignin) with a digestion
291 time of 70 days. Sharma *et al.* [23] reported methane yields on *Ipomoea fistulosa*
292 leaves (25 % lignin) of up to $0.39\text{-}0.43 \text{ m}^3 \text{ CH}_4/\text{kg VS}$ with a digestion time of 8 weeks.
293 Turick *et al.* [19] reported methane yields in the range $0.27\text{-}0.32 \text{ m}^3 \text{ CH}_4/\text{kg VS}$ for
294 various types of woody biomass with a digestion time of 100 days. Only limited
295 investigation has been carried out on the anaerobic digestion of wheatgrass, the
296 substrate used in this study. Our previous study on wheatgrass [24], reported up to 50
297 % removal of TC in batch tests after a digestion time of 150 days. Romano *et al.* [25]
298 investigated the anaerobic digestion of *Jose Tall Wheat Grass* in batch tests (up to 21
299 d), with or without enzyme addition. Without enzyme addition they reported methane
300 yields of up to $0.22 \text{ m}^3 \text{ CH}_4/\text{kg VS}$, which increased up to $0.29 \text{ m}^3 \text{ CH}_4/\text{kg VS}$ with the
301 addition of enzymes. Lalak *et al.* [26] investigated the anaerobic digestion of
302 *Agropyron elongatum* (Tall Wheat Grass) with and without fungi pre-treatment in batch
303 experiments. The methane yield in the test without pre-treatment was $0.13 \text{ m}^3 \text{ CH}_4/\text{kg}$
304 VS , which increased to $0.17 \text{ m}^3 \text{ CH}_4/\text{kg VS}$ with fungi pre-treatment. In comparing our
305 results with the literature, it is important to observe that the different growth stage of
306 the biomass used may influence the results. Indeed, the content of lignin and cellulose
307 changes during germination and growth of the biomass and this can affect its
308 anaerobic biodegradation. There is evidence that the fibre content increases in the
309 initial stages of grass growth [8] and that anaerobic digestion is favoured if the

310 feedstock is harvested in the juvenile crop stages [27]. Our study was carried out with
311 young wheatgrass shoots which are likely to be more biodegradable than more mature
312 biomass. Further study is needed to compare the anaerobic biodegradability of young
313 and more mature wheatgrass plants.

314 As far as the effect of process conditions is concerned, our study is the first to
315 investigate anaerobic digestion of wheatgrass in a semi-continuous process and the
316 effect of the SRT on acidogenic and methanogenic digestion. Furthermore, our study
317 is the first to investigate the use of wheatgrass of the production of chemicals as
318 alternative products to methane.

319 *Effect of inoculum acclimation*

320 The effect of acclimation was investigated by comparing Runs 8 and 9 with Run 2.
321 Acclimation is a period in which microorganisms adapt and adjust to the new
322 environment with a particular substrate, pH and temperature. In Runs 1-7 (presented
323 previously), the semi-continuous mode began on the inoculation day. However, an
324 acclimation of the inoculum in batch mode provides some time for the microorganisms
325 to adapt to the new environment. Figure 5 shows the time profiles in the runs with
326 different acclimation time (Runs 8 and 9) and Figure 6 shows the average values. The
327 initial conditions in the runs with acclimated microorganisms showed high removal of
328 VSS and TC, the removal being higher at the longer acclimation time. This indicates
329 that during the batch acclimation the microorganisms were active and developed the
330 ability to hydrolyse and metabolise the substrate. However, during the semi continuous
331 run the removal of VSS and TC decreased rapidly, and, at steady state, the
332 performance of the acclimated process was only slightly better than for the
333 unacclimated one at steady state condition (Figure 6). A slight increase in the average
334 values of VSS and TC removal and products yield was observed for the fermenter with
335 acclimated inoculum. Whilst the VSS and TC removals were around 12 and 33 %

336 respectively, the VSS and TC removal increased to 15-16% and 39-40% when the
337 inoculum was acclimated respectively. According to Figure 6b the total products yield
338 increased from 13 to 18-20 % gCOD/gCOD_{feed} when acclimation was applied in the
339 fermenters.

340 Acclimation can be an important phase in which the microorganisms can adapt and
341 evolve to the unfamiliar environment and substrate. In this study, considerable grass
342 degradation was observed when the inoculum was acclimated for long periods under
343 batch conditions. However, when the fermenter was set at semi-continuous mode, the
344 substrate removal and products yields were only slightly better than observed in the
345 reactors without inoculum acclimation. The low SRT applied in these experiments (2
346 d) is likely the main reasons for the decrease in performance of the microorganisms.
347 This indicates that the acclimated microorganisms were slow growers that were
348 washed out at the relatively short SRT applied in these runs.

349 Possibly, the combination of batch inoculum acclimation and continuous processing
350 at SRT values longer than 2 d might give a better performance in terms of feedstock
351 removal and yield of liquid phase products. Very few studies tried to use inoculum
352 acclimation to improve the substrate degradation/hydrolysis [28] but their studies were
353 usually operated in batch condition with no investigation of the ability of the acclimated
354 microorganisms to be retained in continuous flow processes.

355 *Effect of temperature*

356 The effect of temperature was investigated by comparing Runs 10-12 with Run 2.
357 Average values at steady state of these runs are shown in Figure 7. The general trend
358 indicates a slight decrease in VSS and TC removal at 50-60 °C and a more significant
359 decrease at 70 °C. Similar observations can be made for the generation of liquid-
360 phase products. No removal of the TCOD was observed in these runs. Regarding the

361 products distribution, acetate was the main product in all the runs, whilst propionate
362 decreased with increasing temperature.

363 Although some studies reported that thermophilic conditions ($T > 40\text{ }^{\circ}\text{C}$) favours the
364 hydrolysis and substrate removal [21, 29, 30] these studies were usually carried out
365 with long acclimation periods, ranging from several months to one year, to each
366 temperature condition, whilst in our study the semi-continuous runs were started
367 immediately after inoculation, with no acclimation to the different temperatures. The
368 effect of acclimation time at the different temperatures deserves further investigation.

369 *Effect of buffer capacity*

370 The effect of buffer capacity was investigated by comparing Run 13 with Run 2. The
371 average values of the pH, VSS and TC removal and products yield at steady state
372 condition in the runs at 20 and 200 mM phosphate buffer are shown in Figure 8. The
373 main difference between these experimental runs was the pH dropped from 6.8 to 5.0
374 when phosphate buffer concentration was reduced from 200 mM to 20 mM (Figure
375 8a). The consequent difference in pH affected the substrate removal and contributed
376 to a shift in the products distribution. A higher substrate removal and products yield
377 were obtained for the fermenter at 200 mM of phosphate buffer and more neutral pH
378 (Figure 8b). Although acetic acid was still the main fermentation product for both
379 fermenters, the production of propionic acid was more significant for the fermenter with
380 phosphate buffer 200 mM than for the fermenter with phosphate buffer 20 mM (Figure
381 8c). Butyrate and ethanol were obtained at very low concentration in both conditions.
382 The fermentation at high and low phosphate buffer concentration showed that the
383 effect of pH on the fermentation performance is considerable. The strong effect of pH
384 is probably due to the fact that hydrolytic enzymes produced in-situ usually have an
385 optimum activity around neutral and high pH [31, 32]. The lower substrate removal

386 and products yield at acidic pH might also be associated to microbial inhibition due to
387 pH toxicity.

388 The microbial inhibition at low pH is usually associated to the "weak-acid uncoupling"
389 mechanism [33]. The carboxylic acids in their undissociated form at low pH are more
390 permeable to the membrane cell, diffuse passively, and acidify the cytoplasm, causing
391 inhibition of cell metabolism.

392 Regarding the products yield, an increased production of propionate was observed
393 with the higher concentration of phosphate buffer (higher pH). Although no study on
394 effect of pH in anaerobic digestion of grass could be found, several other studies have
395 observed how the fermentation products are distributed in a wide pH range [31, 34].
396 The difference observed in several studies demonstrate that the products distribution
397 is very dependent of the feedstock nature and pH simultaneously.

398 Although in this study a phosphate buffer at different concentrations was used, in full
399 scale digesters the use of controlled acid or base addition could be more feasible for
400 pH control. This does not alter the significance of our results on the effect of pH on the
401 process.

402 *General discussion and significance of the results*

403 Whilst most literature studies on the AD of lignocellulosic feedstocks are focussed on
404 the pre-treatment steps, this study shows that AD is a promising technology for the
405 conversion of lignocellulosic substrates, at least for the substrate considered in this
406 study, even without any hydrolytic pre-treatments. In this study, up to 20 % yield (COD
407 basis) of liquid phase products (SRT 8 d) and up to 57 % yield (COD basis) of gas-
408 phase products (SRT 64 d) were obtained.

409 The SRT is the most important parameter that determines the conversion of the
410 substrate and the type of products obtained (methane or liquid phase products). At

411 low SRT the conversion of the feedstock is limited and organic acids are the main
412 fermentation products. On the other hand, at high SRT the conversion of the feedstock
413 is much higher but almost no liquid phase products are present in the effluent due to
414 high methanogenic activity.

415 If the aim of the AD process is convert the lignocellulosic feedstock into methane, then
416 long SRT values should be applied. However, in AD processes with suspended
417 biomass and without biomass recycle the SRT and HRT coincide, therefore high
418 values of the SRT correspond to large vessel volumes and, consequently, the capital
419 expenditure becomes higher and the volumetric methane production becomes lower.
420 In order to reduce reactor volumes, processes with long SRT and short HRT could be
421 considered. UASB and packed bed reactors allow decoupling the SRT from the HRT,
422 however these processes only tolerate low concentrations of suspended solids in the
423 feed (typically less than 1 %). The anaerobic gas lift reactor (AGR) has been proposed
424 for the decoupling of SRT and HRT in systems with high concentrations of suspended
425 solids in the feed [35].

426 If the aim of the AD process is to produce liquid phase products (organic acids and
427 ethanol), then conditions need to be found that give high conversion of the feedstock,
428 high products concentration and no methane production. Acclimation of the
429 microorganisms to the feedstock in batch before the start of continuous processing at
430 low SRT can possibly improve the substrate hydrolysis. However, this study showed
431 that, even though microorganisms in batch conditions can achieve high degradation
432 of the feedstock, when the conditions are switched to continuous with low SRT, the
433 biodegradation of the feedstock decreases significantly. Possibly, batch acclimation
434 coupled with longer SRT could give better process performance and avoid washing
435 out the slow-growing microorganisms that can hydrolyse the lignocellulosic substrates.
436 However, the SRT would still need to be short enough (e.g. 4 or 8 d) to wash out

437 methanogenic microorganisms. As an alternative, continuous acclimation at long SRT
438 followed by gradual reduction in the SRT could be investigated to obtain high
439 conversion of the feedstock and wash-out of the methanogenic microorganisms.

440 Another interesting finding of this study is the long time required to achieve high
441 conversion of the lignocellulosic substrate. In the run at the longest SRT (64 d),
442 microorganisms developed their maximum degradation potential of the feedstock only
443 after about 70 days from the start of the continuous run. This indicates the importance
444 of allowing a long enough acclimation time in the anaerobic digestion of lignocellulosic
445 substrates.

446 This study also shows that, in addition to the SRT, pH and temperature are also
447 important operational variables in determining the conversion of the feedstock and the
448 product yield. Low pH values need to be avoided. Temperatures higher than 40 °C
449 don't give any benefits in process performance and the high temperature of 70 °C
450 causes a significant decrease of feedstock conversion and product yield. However,
451 the effect of pH and temperature has been investigated in this study at the relatively
452 low SRT of 2 d, and further study needs to be done in a wider range of SRT values.

453 Another important aspect that needs further investigation is the concentration of
454 products in the digestate. Clearly, high concentration of products need to be obtained
455 for an economical separation after the AD process. In this study, with a feedstock
456 concentration of 24 gCOD/l, the maximum products concentration obtained was 4.8 g
457 COD/l. In order to increase the product concentration in the effluent, studies with
458 higher concentration of the feedstock are needed.

459 Finally, it is important to observe that the results obtained in this study depend on the
460 feedstock used, the stage at which it was harvested and the pre-treatments used by
461 the manufacturer. In particular this study was carried out with a milled and dried
462 feedstock. Milling reduces the particle size of the feedstock, increasing the surface

463 area available for microbial attachment, the first step in microbial hydrolysis.
464 Therefore, reaction rates with smaller particle sizes of the feedstock are likely to be
465 higher than with larger particle size, if the hydrolysis step is controlling. More research
466 needs to be carried out to investigate the role of the particle size of the lignocellulosic
467 feedstock on the rate of anaerobic digestion. More in general, the effect of the nature
468 of the lignocellulosic feedstocks on their anaerobic digestibility deserves further
469 investigation.
470

471 **Conclusions**

472 This study has shown that high degradation of wheatgrass is possible under anaerobic
473 conditions without pre-treatment. However, stable production of liquid-phase products
474 (SCOAs and ethanol) was only possible at low-medium SRT (up to 8 d), and at this
475 condition the removal of the feedstock was limited (10-15 % removal of the VSS, up
476 to 30% removal of TC), limiting the maximum yield of liquid-phase products to 20 %
477 (COD basis). High SRT values (64 d) gave high conversion of the feedstock, but no
478 liquid phase products were observed and TCOD removal was up to 57 %, indicating
479 that methane production was the main mechanism for substrate removal.

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487

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592 of poultry litter through high rate biomethanation technology: A full scale experience.
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594

595 **Table 1.** Characterisation of the feedstock suspension. TS = total solids; VS = volatile
596 solids; TSS = total suspended solids; VSS = volatile suspended solids; TC = total
597 carbohydrates; SC = soluble carbohydrates; TCOD = total chemical oxygen demand;
598 SCOD = soluble chemical oxygen demand.

TS (g/l)	VS (g/l)	TSS (g/l)	VSS (g/l)	TC (g/l)	SC (g/l)	TCOD (g COD/l)	SCOD (g COD/l)	pH
24	19	22	17	10.1	1.7	24	3	6.8

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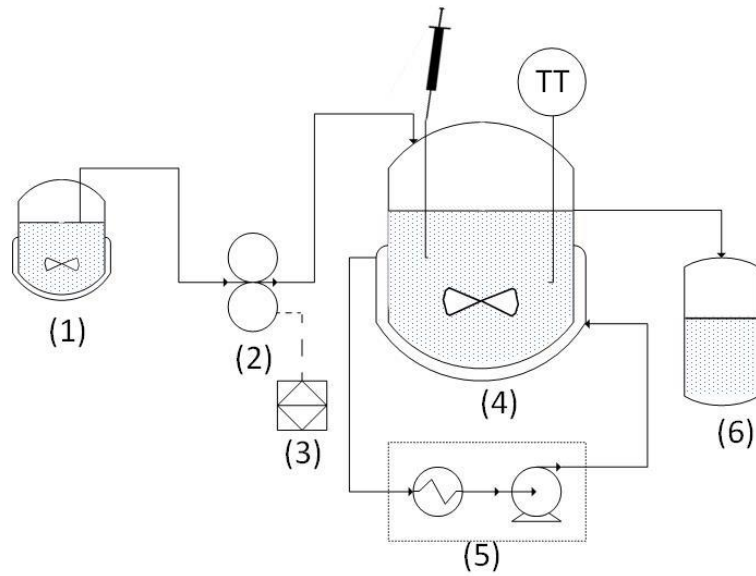
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601 **Table 2.** Summary of experimental conditions in all the runs. OLR= organic load rate.

	SRT (d)	T (°C)	Interval between feeds (d)	Pre- acclimation	Buffer concentration (mM)	OLR (gCOD/l.d)
Run 1	1	40	0.125	No	200	24
Run 2	2	40	0.25	No	200	12
Run 3	4	40	0.5	No	200	6
Run 4	8	40	1	No	200	3
Run 5	16	40	2	No	200	1.5
Run 6	32	40	4	No	200	0.75
Run 7	64	40	8	No	200	0.375
Run 8	2	40	0.25	30 d	200	12
Run 9	2	40	0.25	80 d	200	12
Run 10	2	50	0.25	No	200	12
Run 11	2	60	0.25	No	200	12
Run 12	2	70	0.25	No	200	12
Run 13	2	40	0.25	No	20	12

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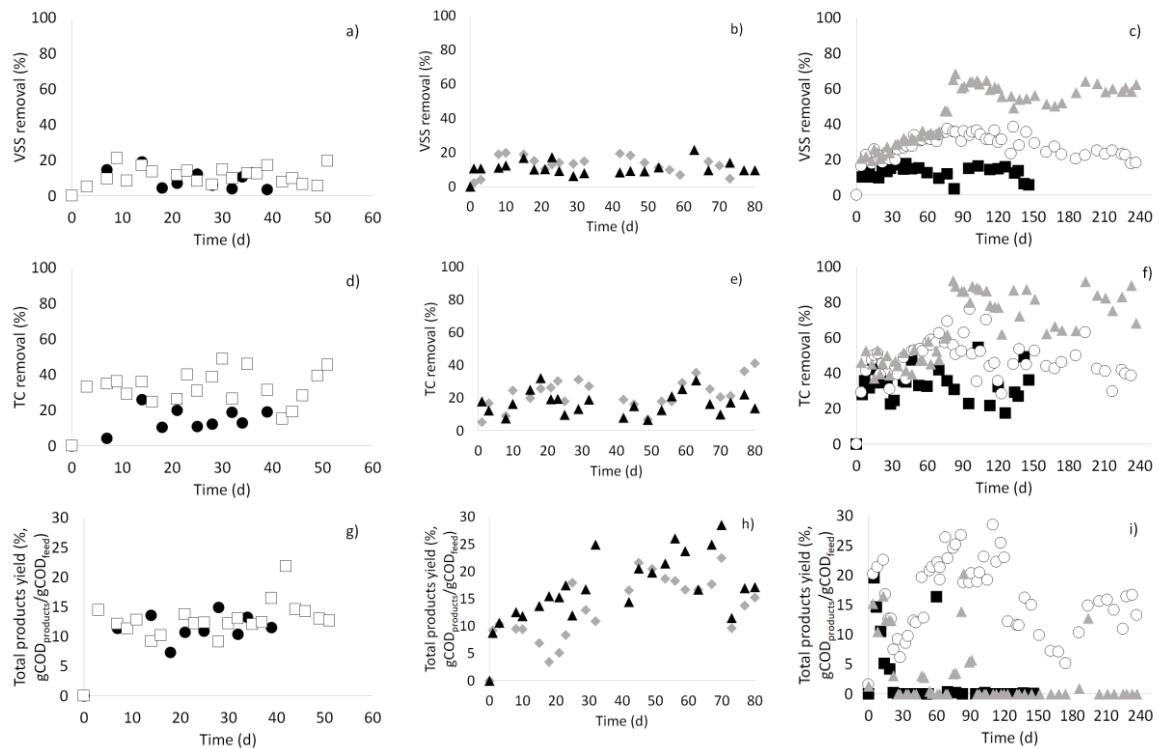
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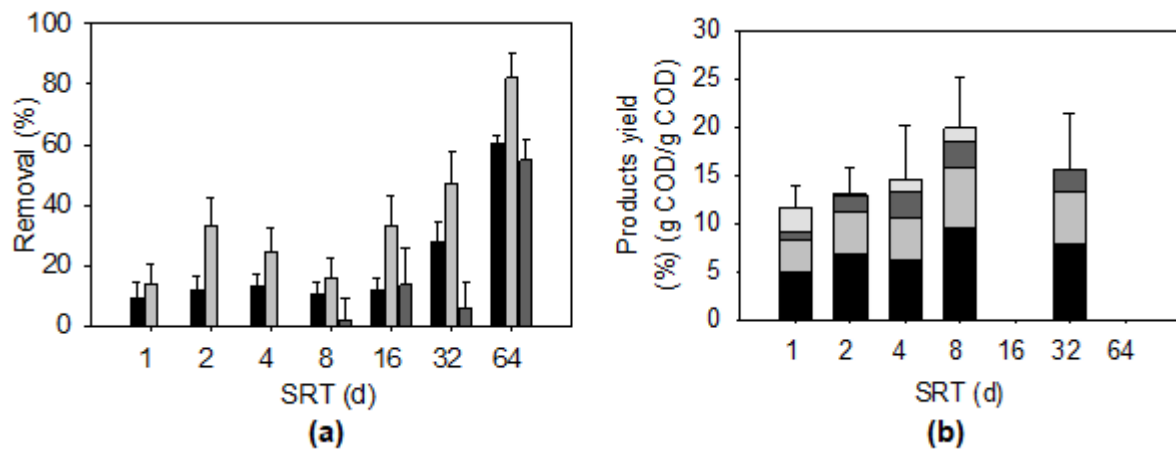
605 **Figure 1.** Scheme of the experimental set-up. (1) Feedstock solution; (2) Peristaltic
606 pump; (3) Programmable power management system; (4) Fermentation glass vessel
607 containing a magnetic stir bar, one sampling tubing and a thermostat; (5) Heating
608 circulator; (6) Effluent

609



611 **Figure 2.** Time profiles for the fermentation runs (Runs 1-7) as a function of the SRT.
 612 (a, b, c): VSS removal; (d, e, f): TC removal; (g, h, i): Total products yield. Figures a),
 613 d) and g) show runs at SRT 1 d (●), 2 d (□); Figures b), e), h) show runs at SRT 4 d
 614 (◆), 8d (▲); Figures c), f), i) show runs at SRT 16 d (■), 32 d (○), 64 d (▲).

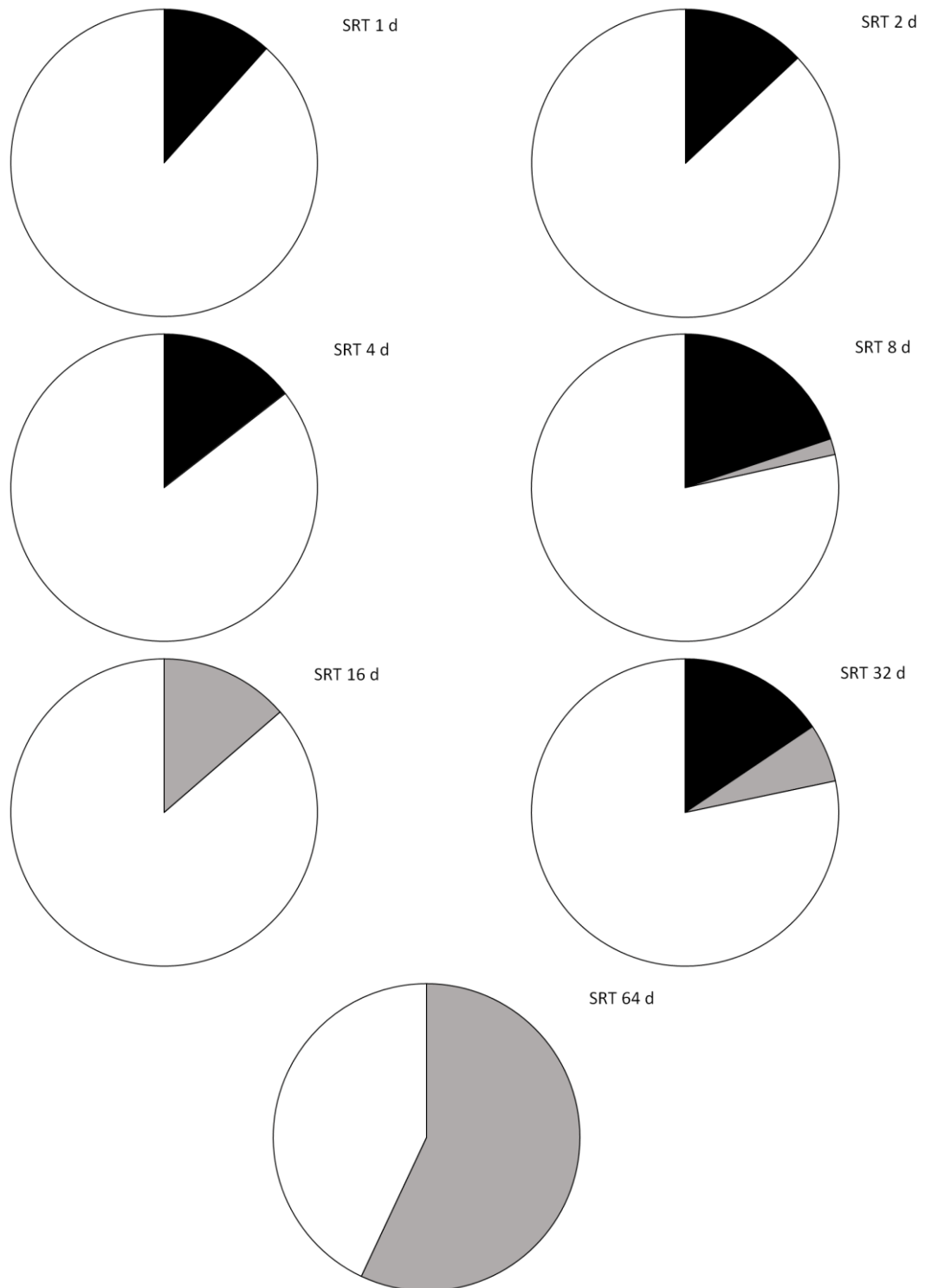
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616

617 **Figure 3.** Average values at steady state for Runs 1-7: (a) Removal from the liquid-
618 solid phase of VSS (■), TC (□) and TCOD (■); (b) Products yield: acetate (■),
619 propionate (□), butyrate (■) and ethanol (□).

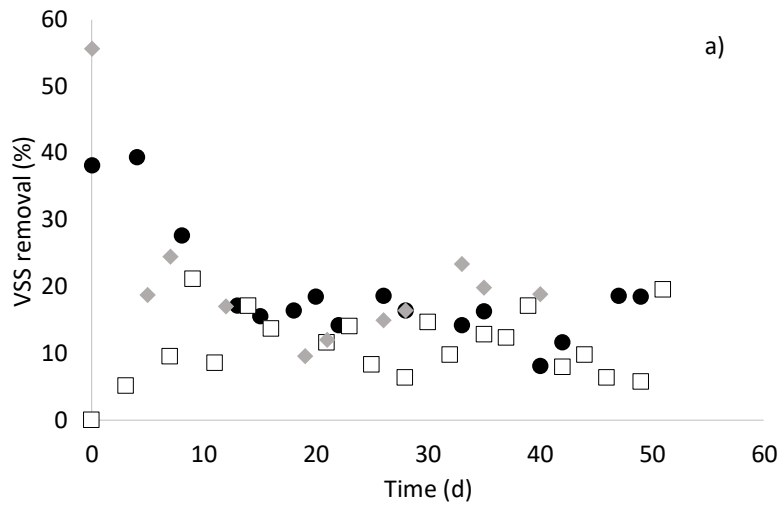
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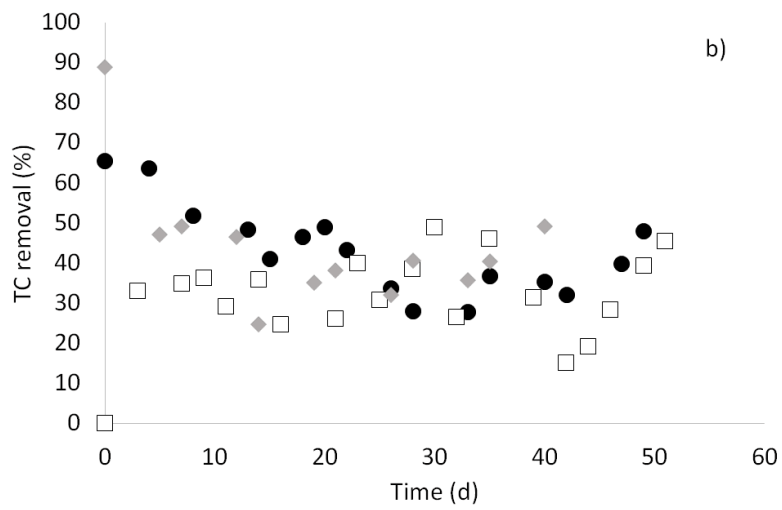
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622 **Figure 4.** Fate of the TCOD of the feed in Runs 1-7 at steady state: ■ , converted to
 623 liquid-phase products; ■ , converted to methane; □ unconverted plus converted to
 624 microorganisms.

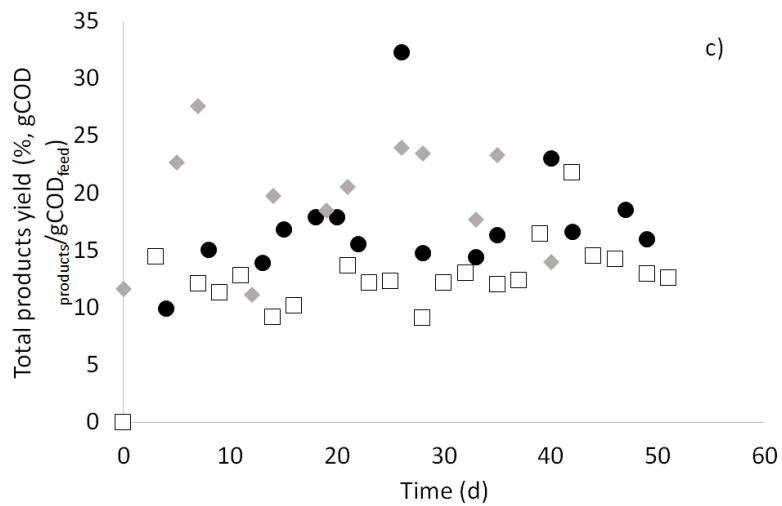
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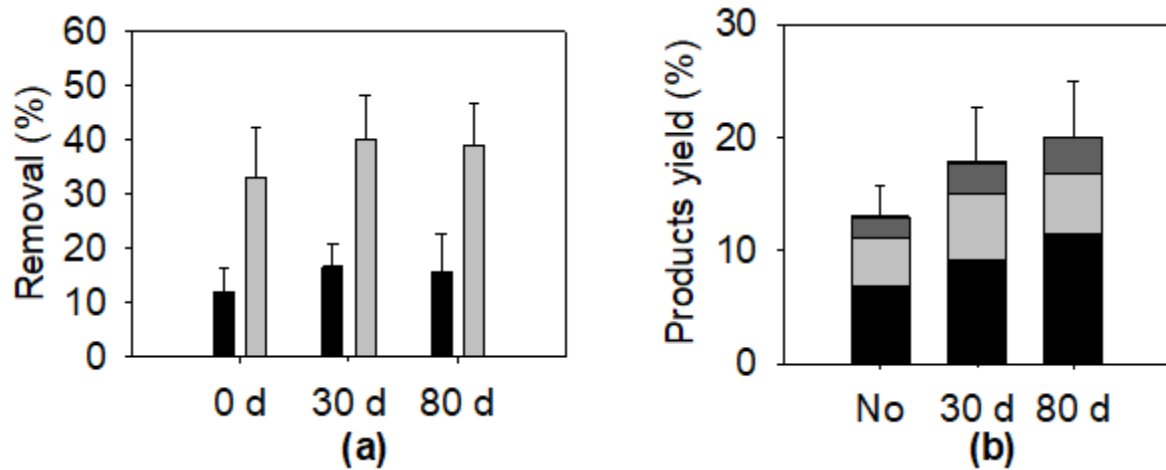


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628

629 **Figure 5.** Time profiles for Runs 8 and 9 compared with Run 2: no acclimation (\square), 30
 630 days of acclimation (\bullet), 80 days of acclimation (\blacklozenge); (a) VSS removal; (b) TC removal
 631 and (c) Total products yield.



632

633 **Figure 6.** Average values at steady state for Runs 8 and 9 compared with Run 2

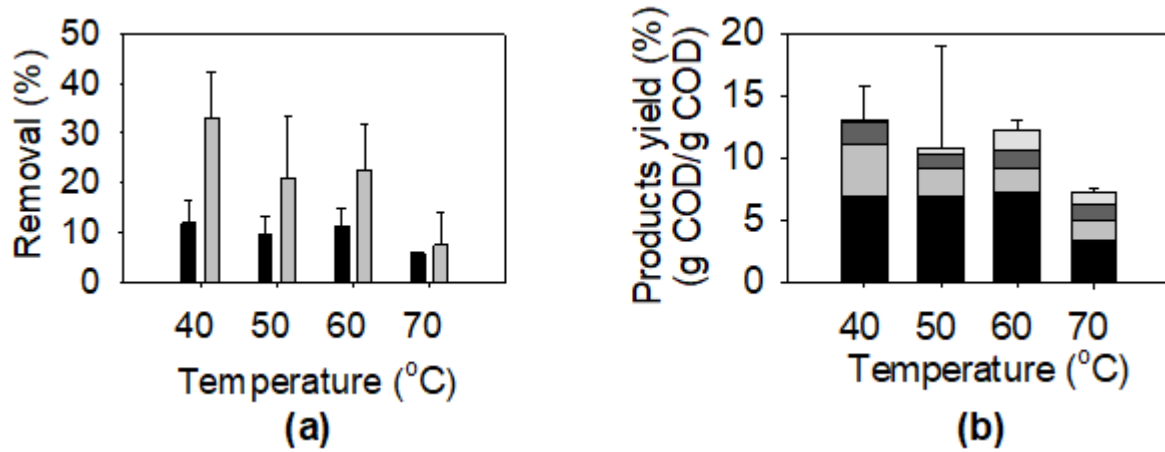
634 (acclimation of 0, 30 and 80 d): (a) Removal from the liquid-solid phase of VSS (■)

635 and TC (□); (b) Products yield: acetate (■), propionate (□), butyrate (■) and

636 ethanol (□).

637

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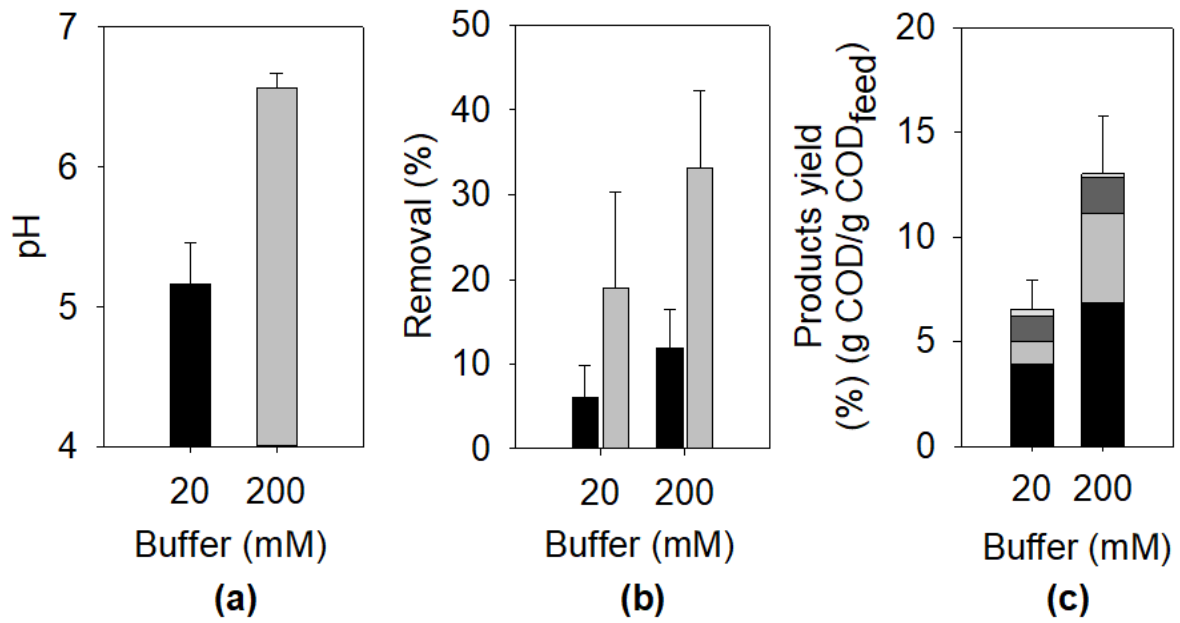
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640 **Figure 7.** Average values at steady state for Runs 10-12 compared with Run 2 (at 40,

641 50, 60 and 70 °C): (a) Removal from the liquid-solid phase of VSS (■) and TC (□);

642 (b) Products yield: acetate (■), propionate (□), butyrate (■) and ethanol (□).

643



644

645 **Figure 8.** Average values at steady state condition for Run 13 compared with Run 2

646 (20 and 200 mM of phosphate buffer): (a) pH; (b) Removal from the liquid-solid phase

647 of VSS (■) and TC (□); (c) Products yield: acetate (■), propionate (□), butyrate

648 (■) and ethanol (□).

649

650